Stereoselective conversion of (+)- $\underline{\text{cis}}$ -2-oxabicyclo[3.3.0] oct-6-en-3-one into methyl $\underline{\text{R}}$ -(+)-7-(3-hydroxy-5-oxo-cyclopentenyl)-5- $\underline{\text{cis}}$ -heptenoate

L. Gruber, I. Tömösközi, E. Major and G. Kovács[₹]

Central Research Institute for Chemistry, The Hungarian Academy of Sciences
*Chinoin Chemical and Pharmaceutical Work Ltd., Budapest, Hungary
(Received in VK 3 September 1974; accepted for publication 12 September 1974)

The antipode of the unsaturated lactone $\underline{1}$ has been recently reported in several papers 1-3 as a starting material for prostanoid syntheses. Since optical resolution of racemic $\underline{1}$ with \mathbf{A} -methylbenzylamine affords both antipodes in a single process each with more than 60% yield, the use of $(+)-\underline{1}$, the enantiomer so far discarded, increases the attrictiveness of $\underline{1}$ in the field of prostanoid syntheses. A highly stereoselective route from $\underline{1}$ to $\underline{6a}$ a well established precursor of PGE₂ and modified analogs is illustrated below.

Hydroxylation was carried out with catalytic amount of $0sO_4$ in the presence of 1.1 mole $NaClO_3$ in water at ambient temperature. Usual work-up and recrystallization from ethyl acetate afforded $2[80-92\%, \text{m.p.}124-5^{\circ}(\text{racemic}), 88-90^{\circ}(+), [<]_{D} +31^{\circ}(\text{c 0.8,MeOH}), ir:3450,1770 cm⁻¹, <math>\frac{\text{H}^1-\text{nmr}^6}{14.9}:4.9(\text{m,l-H}),3.9(\text{m,2-H}),1.7-3.0$ ppm(m,5-H)]. Since only the stereochemistry at C-3 was preserved during further

steps, the opposite configuration at C-1 and C-5 is immaterial. The stereochemistry of hydroxylation is ascertained by known configuration $\frac{4}{100}$ of (-)-6aobtained from (+)-1 of similarly known configuration⁵. Conversion of 2 into dioxolane 3a was made with 2,2-dimethoxypropane and catalytic amount TsOH in benzene [92%, m.p.67.5-68° (racemic), 76-78°(+), $[\alpha]_{p+83.2}$ ° (c 0.55, MeOH), ir:1763, 1392,1378,1075 cm⁻¹, c¹³-nmr⁶: 176.7(C=0),111.4(acetonide-C),86.8,82.1,80.9 $(\underline{C}-0),43.6,36.8,30.8(\underline{CH}_2,\underline{CH}),25.6,23.8 \text{ ppm}(\underline{CH}_3)$. 3a was reduced with 1.4 equivalents i-Bu_AlH in toluene at -60°(2 hrs) to give 3b [85%, ir:3400,1390 1380,1070 cm⁻¹, R_p 0.35(major) and 0.12(minor), silica gel, benzene-methanol, 9:1. Treatment of lactol acetonide with 1.1 equivalent 7 in DMSO5 at room temperature followed by esterification with methyl iodide in DMF or, alternatively, acidification to pH 4.5 with oxalic acid, extraction, and titration with ethereal diazomethane afforded 4 [100%, oil, R. 0.75 silica gel, benzene-MeOH, .9:1, ir: $3600,1745,725 \text{ cm}^{-1}$, H^{1} -nmr: 5.5(m,2-H),4.7(m,2-H),4.0(m,1-H), <math>3.65(s,3-H), 2.3(m,8-H),1.7(m,3-H),1.5(s,3-H),1.3 ppm(s,3-H)] which was further oxidized with 5 equivalent CrO₃-pyridine complex at 150 in methylene chloride to give 5 [96%,oil, ir: 2960,2910,2850,1730 cm⁻¹]. Splitting of the dioxolane ring and dehydration was first attempted in a single process by refluxing 5 with small amount of TsOH in methanol. Chromatography on silica gel 1% MeOH-benzene afforded two products (R_f 0.3 and 0.61, benzene-methanol,9:1) in about the same amount. The more polar substance proved to be $\frac{6a}{a}$ [[α] $_{D}$ +11.2 o (c 1.92,MeOH), ir: 3450,1750,1720,1645 cm⁻¹, uv max(MeOH) 220 nm (ϵ 7.200), H¹-nmr: 7.15 (m,1-H), 5.5(m,2-H), 4.95(m,1-H), 3,68(s,3-H), 2.2(m,8-H), 1.75 ppm (m,2-H)]. Structure 6b was assigned to the less polar material on nmr and ir evidences [ir: $1750,1720,1648 \text{ cm}^{-1}, \text{ H}^{1}-\text{nmr}: 7.2(\text{m},1-\text{H}),5.5(\text{m},2-\text{H}),4.5(\text{m},1-\text{H}),3.65(\text{s},3-\text{H}),$ 3.45(s,3-H), 2.3(m,8-H), 1.7 ppm (m,2-H)]. Careful hydrolysis of the acetonide 5 in 20% sulfuric acid at 0° followed by a dehydration procedure adopted from Sih and coworkers 4 gave 6a free from

References

1. P.A. Grieco and J.J. Reap, J. Org. Chem. 38, 3413 (1973)

appreciable contaminations in 83% yield.

- 2, E.J. Corey, K.C. Nicolaou and D.J. Beames, Tetrahedron Letters 2439 (1974)
 - E.J. Corey and B.B. Snider, J. Org. Chem. 39, 256 (1974)
 - E.J. Corey and R. Noyori, Tetrahedron Letters 311 (1970)
- 3, E.J. Corey and J. Mann, J.Amer. Chem. Soc. 95, 6832 (1973)
- 4, J.B. Heather, R. Sood, P. Price, G.P. Peruzzotti, S.S. Lee, L-F.H. Lee, and C.J. Sih, Tetrahedron Letters 2313 (1973)
- 5, E.J. Corey, N.M. Weinshenker, T.K. Schaaf and W. Huber, J.Amer. Chem. Soc. 91, 5675 (1969)
- 6, 100 MHz nmr data were taken on a Varian spectrometer Model XI-100, values given refer to TMS.